## **More SHC Questions – Answers**

t

10.20 P = 
$$\Delta Q / \Delta t$$
 =  $\Sigma (m c \Delta \theta) / \Delta t$  =  $(C_{kettle} + M_w C_w) \Delta \theta / \Delta t$   
2.25 x 10<sup>3</sup> =  $(450 + 1 x 4200) \Delta \theta / \Delta t$   
 $\Delta \theta / \Delta t$  = 0.484 K s<sup>-1</sup> = 29 K min<sup>-1</sup>

10.25 KE = 
$$\frac{1}{2}$$
 m v<sup>2</sup> = 0.5 x 800 x 20<sup>2</sup> = 160 kJ 10 times = 1600 kJ 20% = 320 kJ  
m c  $\Delta \theta$  = 320 kJ (4 x 1.5) x 420 x  $\Delta \theta$  = 320 kJ  $\Delta \theta$  = 127 °C

10.26 
$$KE_1 = \frac{1}{2} \text{ m v}^2 = 0.5 \times 0.046 \times 40^2 = 36.8 \text{ J}$$
  $KE_2 = \frac{1}{2} \text{ m v}^2 = 0.5 \times 0.046 \times 25^2 = 14.4 \text{ J}$   
 $\Delta KE = 22.4 \text{ J}$  (Note, do not use the speed difference - that would be very wrong!)  
 $\text{m c } \Delta \theta = 22.4 \text{ J}$   $0.046 \times 1600 \times \Delta \theta = 22.4 \text{ J}$   $\Delta \theta = 0.30 \text{ °C}$   
b) Both  $\frac{1}{2} \text{ m v}^2$  and  $\text{m c } \Delta \theta$  involve the same mass, which cancels

c) The temperature of the ball will rise until it reaches a 'dynamic equilibrium', where the rate of gain of energy from hitting the wall equals the rate of loss of energy to the surroundings (i.e. a constant temperature).

10.29	a)	(95 - θ)	(θ - 20)		
	b)	Using m c $\Delta \theta$ :	$1.0 \times 4200 \times (95 - \theta) = 0.70 \times 385 \times (\theta - 20)$		
		Solution gives $\theta$ = 90.5 °C (90.47 °C, so rounds to 90 °C to 2 s.f.)			
	c)	Comment on much higher SHW of water than copper			
SQ 3	a)	mgh = 0.50 x 1	1.30 x 9.81 = 6.38 J	50 times = 319	J
	b)	$m c \Delta \theta = 319$ .	J 0.50 x c x (23 - 18	3) = 319 J	c = 128 J kg <sup>-1</sup> K <sup>-1</sup>
SQ 4	P =	$\underline{m} c \Delta \theta = 0$	.025 x 4200 x (40 - 10) =	3150 W = 3.2 kW	