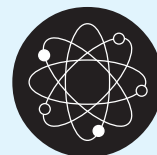
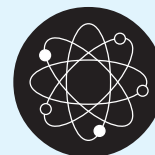


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2. Moles and formulae
3. Moles and equations
4. Shapes of molecules and ions
5. Bonding
6. Structure of elements and compounds
7. Moles and volumetric analysis
8. Energetics I – Hess's law
9. Equilibrium and le Châtelier's principle
10. Kinetics I – rates of Reaction
11. Oxidation and reduction 1
12. Periodicity – trends in period 3
13. Groups 1 and 2
14. Group 7
15. Organic Chemistry I - nomenclature
16. Organic Chemistry II – reactions I
17. Organic Chemistry III – reactions II
18. Applied chemistry

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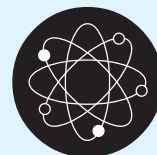
19. The periodic table - period 3
20. The periodic table - group 4
21. Chemical equilibria: K_c
22. Chemical equilibria: K_p
23. How to answer questions on titration calculations
24. Laboratory chemistry: making observations and inferences
25. Acid-Base equilibria I – pH, K_w and K_a
26. Acid Base equilibria II – buffer Solutions, titrations & indicators
27. Organic Chemistry 1 – key concepts and optical isomerism
28. Energetics – lattice enthalpy and Born-Haber cycle
29. Answering questions on Born-Haber cycles
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31. Organic chemistry 2 – halogeno-compounds & Grignard reagents
32. Organic chemistry 3 – carboxylic acids, esters & acyl chlorides
33. Organic chemistry 4 – carbonyl compounds
34. Organic Chemistry 5 – compounds containing nitrogen
35. Answering questions on organic pathways and conversions
36. Laboratory chemistry – continuous practical assessment

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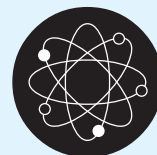
37. Redox equilibria 1: standard electrode potentials & cells
38. Transition metals 1: definitions & properties
39. Reactions of benzene & its compounds
40. Organic reaction mechanisms
41. Answering questions on electrochemical cells
42. Critical analysis of experimental procedures & accuracy
43. Rate equations, orders & constants
44. Rate expression orders & experimental procedures
45. Standard electrode potentials & feasibility of reactions
46. Transition metals 2 - compounds & reactions
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48. Laboratory chemistry - organic techniques
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50. Redox equilibria 3: applications
51. Redox equilibria 4: redox titrations
52. Reactions of functional groups: revision summary
53. Answering questions on organic synthesis
54. Organic analysis 1: infrared spectroscopy

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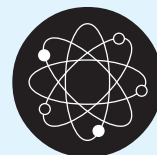
55. Organic analysis II: mass spectrometry
56. Maths for chemists I
57. Answering questions on redox titrations 1
58. Applied organic chemistry
59. Titration calculations: revision summary
60. Laboratory chemistry: summary of organic tests
61. Organic Analysis III - low resolution NMR spectroscopy
62. Answering questions on identifying unknown organic compounds
63. Answering questions on identifying unknown inorganic compounds
64. Acid base III: buffer solutions, pH curves and dibasic acids
65. Calorimetry experiments
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67. Why students lose marks: AS redox questions
68. Periodic table: anomalies of first member of group
69. Revision summary: trends in the periodic table 1
70. Revision summary: trends in the periodic table 2
71. Revision summary: electronegativity, ionisation energies & electron affinities
72. Graphical techniques

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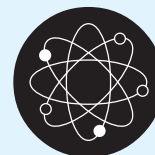
73. Reaction mechanisms - revision summary
74. Ammonia and the Haber process
75. The chemistry of chromium
76. Polymers
77. The importance of hydrogen bonding
78. Recognising, constructing & interpreting redox reactions
79. Catalysts
80. The extraction of aluminium
81. UV and visible spectroscopy
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83. The chemistry of copper
84. Free radical substitution & polymerisation
85. Salt hydrolysis
86. Deprotonation (acid-base reactions)
87. σ and π - bonds and the structure of benzene

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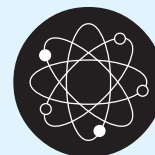
88. Disproportionation
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91. Naming of organic compounds
92. Electrophilic addition
93. Isomerism in organic chemistry
94. Enthalpies of solution
95. Rearranging formulae
96. Relating the properties of crystal structures to structure and bonding
97. Oxidation and reduction in organic chemistry
98. Improve your mark: AS energetics
99. Why students lose marks: A2 acid base calculations
100. Organic functionality and structure – part 1 (AS)
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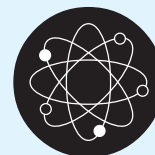
103. Let EMMA do your mole calculations for you
104. A FEW H's will produce any half-equation
105. Logs and powers in chemistry
106. AS chemical bonding: intermolecular bonds
107. AS chemical bonding: intramolecular bonds
108. Amino acids and polypeptides
109. Cracking, reforming and isomerisation
110. $\Delta G = \Delta H - T\Delta S$
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128. Key definitions in AS chemistry
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137. How science works: what do enthalpy of combustion of alcohols data tell us?
138. Industrial uses of transition elements and their compounds
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143. Are biofuels carbon neutral?
144. Catalytic converters
145. Degradable polymers
146. Fuel cells
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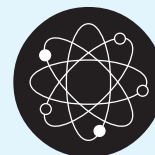
148. Chiral compounds as medicines
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150. How to answer AS exam questions on bonding
151. How to answer AS exam questions on atomic structure
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153. Amines
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162. How to answer questions about structure and bonding

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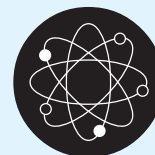
163. Ethanol production
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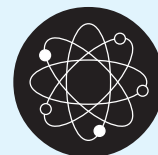
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182. Electronegativity – predicting reaction products
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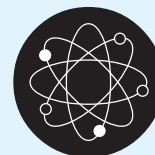
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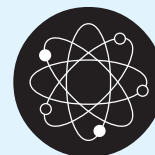
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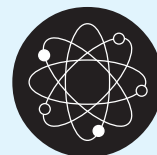
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